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Renewable Energy: Geothermal Solution

- 1. (a) The mass of the air to be heated is : $m_{air} = V \cdot \rho_{air} = 8000m^3 \cdot 1.29 \text{ kg/m}^3 = 10320 \text{ kg}$ The required energy change: $\Delta \dot{E} = m_{air} \cdot c_{air} \cdot \Delta T / \Delta t = 10320 kg \cdot 1000 \frac{J}{kgK} \cdot 7.3 \cdot 10^{-3} K / s = 75336 J / s$ Installed heating capacity: $P_n = 75.3 \text{ kW}$
 - (b) COP of heat pump = 4.2 E = Electrical energy (for pump) $Q_n = \text{Useful heat}$ $Q_u = \text{Ambient heat}$ $Q_n = Q_u + E$

Coefficient of power for heat pump is defined as $\text{COP} = \frac{Q_n}{E} = \frac{P_n}{P_{el}}$ This implies electrical power $P_{el} = 75.3 \text{ kW} / 4.2 = 17.9 \text{ kW}$ Heating power of the probe corresponds to the Q_n and hence $P_n = P_u + P_{el}$ implying $P_u = 75.3 \text{ kW} - 17.9 \text{ kW} = 57.4 \text{ kW}$

- (c) l is the length of the probe. Using P_u from part b we get: $l \cdot 52 \frac{W}{m} = 57.4$ kW implies l = 1104m we also have : $\frac{Q_u}{l} = \frac{Q_n - E}{l} = \frac{Q_n - \frac{Q_n}{COP}}{l}$ Insertion of annual heating demand $Q_n = 135000$ kWh gives: $\frac{Q_u}{l} = 93.5 \frac{kWh}{m} < 110 \frac{kWh}{m}$
- (d) $E = \frac{Q_n}{COP} = 32143$ kWh implies the electricity cost = 32143 kWh · 0.13 Fr./kWh = 4178.59 Fr.
- (e) Oil heating:

Annual heating demand in MJ equal to 135000 kWh = 486000 MJ Volume of oil required is $V_{oil} = \frac{486000 MJ}{42.6 MJ/kg \cdot 0.86 kg/L} = 13266$ litres The price of the oil would be 13266 litres . 0.86 Fr./litre =11408 Fr.

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- (f) CO₂ emission geothermal heat pump : $E \cdot 0.13 \frac{kg}{kWh} = \frac{Q_n}{COP} \cdot 0.13 = 4179 \text{ kg}$ CO₂ emission oil heating : 486000 MJ $\cdot 0.074 \frac{kg}{MJ} = 35964 \text{kg}$ Hence reduction in CO₂ emission is $\frac{(35964 - 4179)}{35964} = 88.4\%$
- 2. (a) Available heat in the geothermal source: $\dot{Q} = \dot{m}c_p(T_{in} T_{out}) = 50*4180*(190-85) = 21.945$ MW

Exergy available in the geothermal source:

$$T_{logmean} = \frac{T_{h,in} - T_{h,out}}{ln \frac{T_{h,in}}{T_{h,out}}} = \frac{463 - 358}{ln \frac{463}{358}} = \frac{105}{0.2572} = 408.25 \text{K}$$

$$Ex_{source} = \dot{Q} * (1 - \frac{T_a}{T_{logmean}}) = 21.945 * (1 - \frac{287}{408.25}) = 21.945 * (1 - 0.703) = 6.52 \text{MW}$$

$$\begin{split} T_{logmean} &= \frac{T_{h,in} - T_{h,out}}{ln \frac{T_{h,in}}{T_{h,out}}} = \frac{333 - 313}{ln \frac{333}{313}} = \frac{20}{0.06194} = 322.9 \text{K} \\ Ex_{heating} &= \dot{Q} * (1 - \frac{T_a}{T_{logmean}}) = 12 * (1 - \frac{287}{322.9}) = 12 * (1 - 0.889) = 1.334 \text{MW} \end{split}$$

Energy Efficiency:

- summer: 3.2 $\mathrm{MW}_e/21.945~\mathrm{MW}$ =14.6%
- winter: $(2.4 \text{ MW}_e + 12 \text{ MW}_{th})/21.945 \text{ MW} = 65.6\%$

Exergy Efficiency:

- summer:
$$\epsilon = \frac{Ex_{electrical} + Ex_{heating}}{Ex_{source}} = \frac{3.2 + 0}{6.52} = 49.1\%$$

- winter: $\epsilon = \frac{Ex_{electrical} + Ex_{heating}}{Ex_{source}} = \frac{2.4 + 1.334}{6.52} = 57.3\%$

(b) The marginal electrical efficiency in winter is the elec. production from the residual heat (21.945 MW - 12 MW DH = 9.945 MW), thus 2.4/9.945=24.1%